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ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl atom) have? 4.

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6. Considering your answer to question 5, how many electrons can fit in the entire 5th energy level? can (AVel is a ha AC5a) 50đ 7.

Write the notation for an electron spinning clockwise in a p sublevel in the 4th energy level.

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3. What is the shape of a p orbital? How many p orbitals are there in a ublevel?  
4.

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