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One pippete ($20\text{cm}^3/25\text{cm}^3$). However, all candidates in a center must use pipettes of the same volume. (c) The usual apparatus and reagents for qualitative work including: (i) Dilute sodium hydroxide solution (ii) Dilute ammonia solution (iii) Dilute hydrochloric acid

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Meter rule-100g and 150g masses-Thread
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pins-Drawing paper-Drawing board-
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physics exam are as follows; Two Retort stands, Drawing board, Fixed Pulley, Drawing papers, Four Thumb pins, Masses (30,20,10 and 5g), Mass =30g covered and labelled W, Mass hanger and Twine.

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TEST: C + 5cm³ of distilled water and shake thoroughly. Divide the solutions into three portions. OBSERVATION: A pale green solution results. INFERENCE: Salt is soluble (2aii) TEST: To the first portion add NaOH solution in drops. OBSERVATION: Dark green gelatinous was formed. INFERENCE: Fe²⁺ present.

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(2aiii) TEST: then in excess

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Measured Dilute sodium hydroxide
solution. Measured Dilute ammonia
solution. Some quantity of Dilute

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hydrochloric acid. A rubber filled with
Distilled water.

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