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hydrogen from water by sodium is an example of a _____ reaction.

_____ activation energy. in order for a chemical reaction to take place, the particles involved must collide with a sufficient amount of _____

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1) calculate moles of each element (if given mass) 2) divide each mole quantity by the smallest mole value from step 1. 3) use whole number values from step 2 as the subscripts to the original equation to write the empirical formula.

Percent Composition.

-the relative amounts of each element in a

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matter that results when
2 or more atoms are
joined by.... A bond
resulting from the
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Choice 24) What is the amount of heat required to raise the temperature of 200.0 g of aluminum by 10°C ? cal (specific heat of aluminum = $0.21 \text{ J/g} \cdot ^{\circ}\text{C}$)
24) A) 4200 cal
B) 420 cal
C) 42 000 cal
D) 420 000 cal
E) none of the above

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Choice Identify the
choice that best
completes the statement
or answers the question.

____ 1. How many
moles of tungsten atoms
are in 4.8×10^{25} atoms

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of tungsten? a. 8.0×10^2
102 moles c. 1.3×10^1
moles b. 8.0×10^1
moles d. 1.3×10^2
moles ____ 2.

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out of a mix 1600s –
Robert Boyle studied
the behavior of gases
and is responsible for
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