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reaction represented
by the (unbalanced)
equation determine
the number of moles
of $\text{NH}_3(\text{g})$ that can be
produced from the

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following: a. 0.20 mol $\text{N}_2(\text{g})$ reacts completely with $\text{H}_2(\text{g})$. b. 0.30 mol $\text{H}_2(\text{g})$ reacts completely with $\text{N}_2(\text{g})$. 2. Considering the reaction represented by the (unbalanced) equation

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isotopes of iron with
mass numbers 54, 56,
57, and 58 There are
two radioactive

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4. What is a photon?

5. Describe Bohr ' s model of the

hydrogen atom. 6.

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